

1 Method for the surface treatment of a titanium dioxide pigment

2 CROSS REFERENCE TO RELATED APPLICATIONS

3 This application claims priority pursuant to Title 35, United States Code,  
4 Section 119(a)- (d) or (f), or 365(b) to the German Patent Application Number DE 102 36  
5 366 filed 8 August, 2002, where the above named application is hereby incorporated  
6 herein by reference in its entirety including incorporated material.

7 FIELD OF THE INVENTION

8 The field of the invention is the field of methods for the surface treatment of  
9 titanium dioxide pigments to produce a titanium dioxide pigment with high greying  
10 resistance and high hiding power, and its use in the production of decorative laminating  
11 paper.

12 BACKGROUND OF THE INVENTION

13 Decorative laminating paper is an element of a decorative thermoset surface, which is used  
14 with preference for finishing furniture surfaces and for laminate floorings. Laminate is the  
15 term used for materials in which wood and paper, for example, are pressed with resin. The  
16 use of special synthetic resins results in extraordinarily high resistance of the laminates to  
17 scratching, impact, chemicals and heat.

18 The use of special-purpose papers (decorative laminating papers) permits the production  
19 of decorative surfaces, where the decorative laminating paper serves not only as facing  
20 paper for unattractive wood material surfaces, but also as a carrier for the synthetic resin.  
21 The requirements imposed on decorative laminating paper include, for example, hiding  
22 power (opacity), light-fastness (greying resistance), colour-fastness, wet strength,  
23 impregnability and printability.

24 In principle, a pigment based on titanium dioxide is eminently suitable for achieving the

1 necessary opacity of the decorative laminating paper. As a rule, a titanium dioxide pigment  
2 or a titanium dioxide pigment suspension is mixed with a fibre suspension during paper  
3 production. The interactions between the individual components (fibres, pigment, water)  
4 contribute to formation of the paper sheet and determine the retention of the pigment. The  
5 term retention refers to the retention of all inorganic substances in the paper during production.  
6 In addition to the pigment and fibres used as feedstock, auxiliaries and additives are  
7 generally also used. These may affect the mechanisms of interaction between the fibres,  
8 the pigment and the water.

9 A number of titanium dioxide pigments exists for applications in decorative laminating  
10 paper. Alongside the most important properties, such as retention and opacity (hiding  
11 power), the greying resistance also plays a decisive role.

12 It is generally known that titanium dioxide is photochemically active. When exposed to UV  
13 radiation in the presence of moisture, decorative laminating paper pigmented with titanium  
14 dioxide displays increasing greying. To avoid this problem, the surface of the pigments is  
15 treated with various substances, for instance with  $\text{Al}_2\text{O}_3$  aquate and a colourless metal  
16 phosphate (US 3 926 660), with zinc phosphate (US 5 114 486), with cerium phosphate  
17 and aluminium phosphate (GB 2 042 573), or only with aluminium phosphate (EP 0 753  
18 546 A2). DE 15 92 873 describes a method for improving the light-fastness of pigments,  
19 where calcination at 600 °C is performed following coating with magnesium silicate.

20  $\text{TiO}_2$  pigments with improved retention properties, having a cores coated with consecutive  
21 layers of aluminium oxide phosphate, aluminium oxide and magnesium oxide, are  
22 presented in EP 0 713 904 B1, US 5,665,366 and US 5,942,281.

23 A  $\text{TiO}_2$  pigment, having a core coated with consecutive layers of zirconium hydroxide or  
24 oxyhydroxide, titanium hydroxide or oxyhydroxide, and co-precipitated phosphate and silica, and  
25 finally a layer of aluminum oxyhydroxide and magnesium oxide, is shown to protect the organic  
26 pigment binder from light and to decrease the loss of gloss in paint in US 6,200,375.

1 All of the preceding publications, patents, and patent applications are hereby included in  
2 their entirety in this application.

### 3 OBJECTS OF THE INVENTION

4 It is an object of the invention to produce a method capable of producing pigments  
5 with high hiding power and simultaneously high greying resistance.

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8 laminating papers.

### 9 SUMMARY OF THE INVENTION

10 The steps of the most preferred method of the invention are:

- 11 a) Addition of a phosphorus compound to an aqueous suspension of titanium dioxide base  
12 material;
- 13 b) Addition of a titanium compound;
- 14 c) Addition of an aluminium compound;
- 15 d) Adjustment of the pH value of the suspension to a pH value of 8 to 10, preferably 8.5  
16 to 9.5;
- 17 e) Addition of a magnesium compound;
- 18 f) Stabilisation of the pH value of the suspension in the range from 8 to 10, preferably 8.5  
19 to 9.5; and
- 20 g) Separation of the TiO<sub>2</sub> pigment by filtration, followed by washing, drying and milling of  
21 the pigment.

### 23 DETAILED DESCRIPTION OF THE INVENTION

24 Using pigments with improved greying resistance produced by prior art methods generally  
25 reduces retention and/or opacity in laminates. Prior art improvements in opacity are

1 accompanied by a deterioration in greying resistance.

2 Therefore, the object of the present invention is to offer a method capable of producing  
3 pigments with high hiding power and simultaneously high greying resistance for use in  
4 decorative laminating papers.

5 The object is solved by subjecting a titanium dioxide pigment to a surface treatment  
6 process characterised by the following steps:

- 7 a) Preparation of an aqueous suspension of titanium dioxide base material,
- 8 b) Addition of a phosphorus compound,
- 9 c) Addition of a titanium compound,
- 10 d) Addition of an aluminium compound,
- 11 e) Adjustment of the pH value of the suspension to a pH value of 8 to 10, preferably 8.5  
12 to 9.5,
- 13 f) Addition of a magnesium compound,
- 14 g) Stabilisation of the pH value of the suspension in the range from 8 to 10, preferably 8.5  
15 to 9.5,
- 16 h) Separation of the  $\text{TiO}_2$  pigment by filtration, followed by washing, drying and milling of  
17 the pigment.

18 Other advantageous versions of the method are described in the dependent claims of the  
19 invention.

20 The object of the invention is, therefore, a method for the surface treatment of titanium  
21 dioxide pigments that results in pigments with high hiding power and high retention with  
22 simultaneously high greying resistance, as well as a pigment with these properties and the  
23 use of this pigment in the production of decorative laminating paper.

24 Surprisingly, it was found that the addition of a titanium compound during the inorganic  
25 surface treatment of the pigment with phosphorus and aluminium compounds is capable

1 of achieving both high opacity and very good greying resistance. No zirconium compounds  
2 were added to the suspension to achieve this result.

3 The surface treatment process is based on  $\text{TiO}_2$  base material, preferably produced by the  
4 chloride process. The term  $\text{TiO}_2$  base material refers to the raw  $\text{TiO}_2$  pigment prior to post-  
5 treatment. The base material can first be milled, for example in a wet-milling process. A  
6 dispersant is preferably added during wet-milling. The milled base material is used to  
7 prepare an aqueous suspension. This suspension can be basic or acidic and is preferably  
8 basic with a pH value of 9 to 11. The method is performed at a temperature of less than  
9  $70^\circ\text{C}$ , preferably at 55 to  $65^\circ\text{C}$ .

10 First, a phosphorus compound is added to the suspension in a quantity of 0.4 to 6.0% by  
11 weight, preferably 1.0 to 4.0% by weight, calculated as  $\text{P}_2\text{O}_5$ , referred to  $\text{TiO}_2$  base  
12 material. Particularly good results are obtained with  $\text{P}_2\text{O}_5$  contents of 1.6 to 2.8% by weight,  
13 referred to the base material. Other suitable phosphorus compounds are preferably  
14 inorganic phosphorus compounds, such as alkali phosphates, ammonium phosphate,  
15 polyphosphates, phosphoric acid or, where appropriate, mixtures of these compounds.  
16 Other phosphorus compounds can, however, also be used.

17 Second, a titanium compound is added, e.g. titanyl sulphate, titanyl chloride or another  
18 hydrolysable titanium compound, or mixtures of these compounds. The quantity of titanium  
19 compound added is 0.1 to 3.0% by weight, preferably 0.1 to 1.5% by weight, and  
20 particularly 0.1 to 1.0% by weight, calculated as  $\text{TiO}_2$  referred to  $\text{TiO}_2$  base material in  
21 the suspension.

22 Third, an aluminium compound of acidic or basic character is preferably subsequently  
23 added to the suspension. Particularly suitable as an acidic aluminium compound is  
24 aluminium sulphate, although this is not to be taken as a restriction. Suitable alkaline  
25 aluminium compounds include sodium aluminate, alkaline aluminium chloride, alkaline  
26 aluminium nitrate or other alkaline aluminium salts, or mixtures of these compounds.

1 The suspension will customarily be stirred for about 30 minutes following each addition, in  
2 order to achieve homogenisation. It is, however, also possible to add the titanium  
3 compound and the aluminium compound simultaneously.

4 In a preferred embodiment of the method, an acid or a base, or a second aluminium  
5 compound, is added in parallel with the aluminium compound, in order to maintain a  
6 constant pH value in the range from 2 to 10, preferably in the range from 4 to 9 and  
7 particularly in the range from 6 to 8. In a particularly advantageous version of the method,  
8 the pH value is controlled by the balanced, parallel addition of sodium aluminate and HCl.  
9 A further procedure consists in keeping the pH value constant by means of the controlled  
10 addition of aluminium sulphate and sodium aluminate.

11 The suspension is subsequently adjusted to a pH value of 8 to 10, preferably 8.5 to 9.5.  
12 A person skilled in the art adjusts the pH value in the customary manner with the help of  
13 appropriate acidic or alkaline compounds. The alkalis used for this purpose include, for  
14 example, alkaline aluminium salts, such as sodium aluminate, alkaline aluminium chloride  
15 or alkaline aluminium nitrate, or bases, such as sodium hydroxide solution or ammonia, or  
16 a combination of these alkalis.

17 The total quantity of aluminium added to the suspension by way of the various aluminium  
18 compounds is 2.0 to 7.5% by weight, preferably 3.5 to 7.5% by weight, calculated as  $\text{Al}_2\text{O}_3$ ,  
19 referred to  $\text{TiO}_2$  base material.

20 0.1 to 1% by weight, preferably 0.2 to 0.5% by weight, of a magnesium compound is then  
21 added, calculated as MgO and referred to  $\text{TiO}_2$  base material. Suitable for use as the  
22 magnesium compound are water-soluble magnesium salts, such as magnesium sulphate,  
23 magnesium chloride and other magnesium salts, as well as mixtures of these compounds.  
24 The pH value should be maintained at 8 to 10, preferably at 8.5 to 9.5, and most preferably  
25 approximately 8 with the help of appropriate alkaline media, if necessary.

1 The post-treated TiO<sub>2</sub> pigment is subsequently separated from the suspension by filtration,  
2 and the resultant filter cake is washed.

3 To further improve the greying resistance, the pigment can additionally be treated with  
4 nitrate at a concentration of up to 1.0% by weight in the finished pigment.

5 Moreover, the final pH value of the pigment may be set by adding a suitable pH modifying  
6 substance such as an acid, a base, an acid salt, or a basic salt, or a combination of  
7 suitable substances. The pH value is controlled by means of the degree of acidity and the  
8 added quantity of the substance. In principle, all compounds may be used which do not  
9 impair the optical pigment properties, which are temperature resistant during the final  
10 pigment drying or pigment milling and which can be added to the filter paste, into the dryer  
11 or during steam jet milling. For instance, acids like sulphuric acid, nitric acid, hydrochloric  
12 acid or citric acid or acidic salts like chlorides, sulfates or the like are suitable if they comply  
13 to the conditions specified.

14 Nitrate compounds are particularly suitable. By utilizing sodium nitrate the final pH value  
15 comes to more than 9. Yet, the decrease of the pH value can be achieved by the use of  
16 acidic nitrate compounds or a combination of acidic and non-acidic nitrate compounds as  
17 for instance aluminium nitrate, a combination of aluminium nitrate and sodium nitrate, a  
18 combination of aluminium nitrate and nitric acid and so forth. For example, the addition of  
19 aluminium nitrate in a quantity of 0,4% by weight calculated as NO<sub>3</sub> in the filter paste  
20 results in a lowered final pH value of from 7.5 to 8.5, preferably approximately 8.

21 Finally, the pigment is dried and milled.

22 In comparison with the reference pigments, the pigment produced according to this method  
23 displays improved hiding power and improved greying resistance in the laminate, as well  
24 as good retention, and is outstandingly suitable for use in decorative laminating paper.

1     Examples:

2     An example of the invention is described below. Unless otherwise stated, the quantity data  
3     refer to  $\text{TiO}_2$  base material in the suspension.

4     Example 1

5     After sand-milling, a suspension of titanium dioxide from the chloride process with a  $\text{TiO}_2$   
6     concentration of 400 g/l is adjusted to a pH value of 10 with NaOH at 60 °C. 2.4% by  
7     weight  $\text{P}_2\text{O}_5$  in the form of disodium hydrogenphosphate solution is added to the  
8     suspension while stirring. The solution is added over a period of 60 minutes. After further  
9     stirring for 30 minutes, 0.2% by weight  $\text{TiO}_2$  in the form of titanyl sulphate solution is then  
10    added. This is followed by further stirring for 30 minutes. In the next step, 2.7% by weight  
11     $\text{Al}_2\text{O}_3$  in the form of acidic aluminium sulphate solution is mixed into the suspension within  
12    30 minutes. After stirring for 30 minutes, the acidic suspension is set to a pH value of 9.0  
13    with the help of an alkaline sodium aluminate solution in a quantity of 3.7% by weight,  
14    calculated as  $\text{Al}_2\text{O}_3$ . The solution is added over a period of 40 minutes. After stirring for 30  
15    minutes, 0.5% by weight MgO in the form of a magnesium sulphate solution is added. After  
16    further stirring for 30 minutes the suspension is set to a pH value of 9 with NaOH.

17    After being stirred for a further 2 hours, the post-treated  $\text{TiO}_2$  suspension is filtered and  
18    washed. Following the addition of 0.25% by weight  $\text{NO}_3$  in the form of  $\text{NaNO}_3$ , referred to  
19     $\text{TiO}_2$  pigment, the washed filter paste is dried in a spray drier and subsequently steam-  
20    milled.

21    Comparative example 1

22    The pigment is produced in a manner comparable to that described in Example 1, except  
23    that titanyl sulphate and magnesium sulphate are not components of the post-treatment.  
24    While stirring, 2.4% by weight  $\text{P}_2\text{O}_5$  in the form of disodium hydrogenphosphate solution is  
25    added to the sand-milled  $\text{TiO}_2$  suspension (400 g/l  $\text{TiO}_2$ ), which has a temperature of 60 °C  
26    and a pH value of 10. In the next step, 3.0% by weight  $\text{Al}_2\text{O}_3$  is mixed into the suspension



in the form of acidic aluminium sulphate solution. The acidic suspension is set to a pH value of 7.2 with the help of an alkaline sodium aluminate solution in a quantity of 3.4% by weight, calculated as  $\text{Al}_2\text{O}_3$ . The further processing steps (filtration, washing, nitrate treatment, drying, milling) are the same as in Example 1.

#### Comparative example 2

The pigment is produced in a manner comparable to that described in Example 1, except that titanyl sulphate is not a component of the post-treatment.

While stirring, 2.4% by weight  $\text{P}_2\text{O}_5$  in the form of disodium hydrogenphosphate solution is added to the sand-milled  $\text{TiO}_2$  suspension (400 g/l  $\text{TiO}_2$ ), which has a temperature of 60 °C and a pH value of 10. In the next step, 2.6% by weight  $\text{Al}_2\text{O}_3$  is mixed into the suspension in the form of acidic aluminium sulphate solution. The acidic suspension is set to a pH value of 9.2 with the help of an alkaline sodium aluminate solution in a quantity of 3.0% by weight, calculated as  $\text{Al}_2\text{O}_3$ . This is followed by the addition of 0.5% by weight  $\text{MgO}$  in the form of magnesium sulphate solution.  $\text{NaOH}$  is used to set a pH value of 9. The further processing steps are the same as in Example 1 and Comparative example 1.

#### Test methods

The titanium dioxide pigments produced as described above were incorporated into decorative laminating paper based on melamine resin and subsequently tested with regard to their optical properties and greying resistance in pressed laminates. To this end, the titanium dioxide pigment to be tested was incorporated into cellulose, and sheets with a sheet weight of roughly 100 g/m<sup>2</sup> and a  $\text{TiO}_2$  content of about 40% by mass were produced.

##### a) Laminate production (laboratory scale)

A 36.5% aqueous pigment suspension made of 146 g titanium dioxide pigment and 254 g tap water is prepared. Testing is based on 30 g pulp (oven-dry). The corresponding quantity of pigment suspension is adapted to the retention and the required ash content,

1 40 ± 1% in this case, and the grammage, 100 ± 1 g/m<sup>2</sup> in this case. A person skilled in the  
2 art is familiar with the procedure and the auxiliaries used.

3 The ash content (titanium dioxide content) of a sheet and the retention of the pigment are  
4 subsequently determined. The ash content is determined by incinerating a defined weight  
5 of the produced paper in a rapid incinerator at 900 °C. The TiO<sub>2</sub> content by mass  
6 (equivalent to the ash content) can be calculated by weighing the residue.

7 The retention is defined as the capacity to retain all inorganic substances in the sheet of  
8 paper on the wire screen of the paper-making machine. The "one-pass retention" indicates  
9 the percentage retained during a single feeding step to the paper-making machine. The  
10 ash content in percent referred to the percentage by mass of the pigment used relative to  
11 the total solids in the suspension yields the retention.

12 The further processing of the paper encompasses its impregnation and pressing into  
13 laminates. The sheet to be impregnated with resin is immersed in a resin solution and pre-  
14 condensed for 25 seconds at 130 °C in a recirculating-air drying oven. Impregnation is  
15 performed a second time in similar manner, where the dwell time in the drying oven is 110  
16 seconds. The sheet has a residual moisture content of 4 to 6% by weight. The condensed  
17 sheets are combined into stacks with phenolic resin-impregnated core papers, and white  
18 and black underlay paper.

19 The laminate structure used for the test comprised 9 layers: décor sheet, décor sheet, core  
20 paper, core paper, black underlay, core paper, core paper, black/white underlay, décor  
21 sheet.

22 The stacks are pressed for 300 seconds with the help of a Wickert Type 2742 laminating  
23 press at a temperature of 140 °C and a pressure of 90 bar.

## 24 b) Testing

1 The optical properties and the greying resistance of the laminates were measured using  
2 commercially available equipment (spectrophotometer, Xenotest weathering machine).

3 In order to assess the optical properties of laminates, the optical values (CIELAB  $L^*$ ,  $a^*$ ,  
4  $b^*$ ) to DIN 6174 are measured with the help of the ELREPHO<sup>®</sup> 3000 colorimeter over  
5 white and black underlay. The opacity is a measure of the light transmission of the paper.  
6 The following parameters were selected as a measure of the opacity of the laminates:  
7 CIELAB  $L^*_{black}$ , the brightness of the laminates measured over black underlay paper, and  
8 the opacity value  $L [\%] = Y_{black}/Y_{white} \times 100$ , determined from the Y-value measured over  
9 black underlay paper ( $Y_{black}$ ) and the Y-value measured over white underlay paper ( $Y_{white}$ ).  
10 The values are measured using a spectrophotometer (ELREPHO<sup>®</sup> 3000).

11 To assess the greying resistance (light-fastness) of the titanium dioxide pigments or  
12 titanium dioxide pigment blends, the corresponding laminate samples are exposed in a  
13 XENOTEST<sup>®</sup> 150S. The side of the laminate on which two papers are laminated together  
14 is measured for the assessment. The CIELAB  $L^*$ ,  $a^*$  and  $b^*$  optical values to DIN 6174 are  
15 measured before and after 96 hours of exposure in the XENOTEST<sup>®</sup> 150S. The light  
16 source is a xenon-arc lamp. The temperature inside the device is  $23 \pm 3$  °C, the relative  
17 humidity being  $65 \pm 5\%$ . The samples are rotated during the exposure cycle. Both  $\Delta L^* =$   
18  $L^*_{before} - L^*_{after}$  and  $\Delta E^* = ((\Delta L^*)^2 + (\Delta a^*)^2 + (\Delta b^*)^2)^{1/2}$  are stated as a measure of the greying  
19 resistance.

## 20 Test results:

21 The test results for the laminates produced using the pigment according to the invention  
22 from the Example 1 and the pigments from Comparative examples 1 and 2 are  
23 summarised in Table I. All three examples and comparative examples are set to the same  
24 ash content.

25 It can be seen that the laminate produced using the pigment according to the invention  
26 (Example 1) is characterised by both high opacity ( $L^*_{black}$  and L) and high greying resistance

1 ( $\Delta L^*$  and  $\Delta E^*$ ). In contrast, the laminates produced using the two reference pigments 1 and  
2 2 display significantly lower values for either opacity (Comparative example 1) or greying  
3 resistance (Comparative example 2). In addition, the retention of the paper produced using  
4 the pigment according to the invention was improved relative to Comparative example 2.

5 Obviously, many modifications and variations of the present invention are possible in light  
6 of the above teachings. It is therefore to be understood that, within the scope of the  
7 appended claims, the invention may be practiced otherwise than as specifically described.